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Answer six questions choosing two from Section A, two from Section B and two from Section C. Section - A (Answer any two questions) Question 2. (a) (i) A solution containing 0.5 g of KCl dissolves in 100 g of water and freezes at -0.24°C . Calculate the degree of dissociation of the salt.

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(K f for water = 1.86°C), Atomic weights [K = 39, Cl = 35.5] [3]

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(c) Give one word or phrase for the following: The ratio of the mass of a certain volume of gas to

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the mass of an equal volume of hydrogen under the same conditions of temperature and pressure. Formation of ions from molecules. Electrolytic deposition of a superior metal on a baser metal.

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